Tetrahedron 66 (2010) 9493-9496

Contents lists available at ScienceDirect

Tetrahedron

journal homepage: www.elsevier.com/locate/tet

Regio and diastereoselective synthesis of functionalized 2,3-dihydrofuro[3,2-*c*]coumarins via a one-pot three-component reaction

Elisa Altieri, Massimiliano Cordaro, Giovanni Grassi, Francesco Risitano*, Angela Scala

Dipartimento di Chimica Organica e Biologica, Università degli Studi di Messina, Vill. S. Agata, I-98166 Messina, Italy

ARTICLE INFO

Article history: Received 27 July 2010 Received in revised form 23 September 2010 Accepted 11 October 2010 Available online 15 October 2010

Keywords: Multicomponent reaction Cyclization Diastereoselectivity 4-Hydroxycoumarin

ABSTRACT

An efficient and straightforward synthesis of furo[3,2-*c*]coumarins via the one-pot three-component condensation of aromatic aldehydes, 4-hydroxycoumarin and α -chloroketones in refluxing *n*-propanol is described. Pyridine or a mixture of AcOH and AcONH₄ was used as a basic catalyst.

© 2010 Elsevier Ltd. All rights reserved.

Tetrahedror

1. Introduction

Multicomponent reactions (MCRs)¹ involving domino processes are increasingly emerging as a good way of rapidly and efficiently generating complex molecules with potential biological properties.² Our interest in these reactions to synthesize spiro or fused five- and six-membered bicyclic rings is well documented.³ Recently,⁴ we have successfully introduced the use of phenacyl chloride in new MCRs employing aromatic aldehydes as electrophiles and azol-5-ones as pronucleophiles with AcOH/AcONH₄ performing the dual function of catalyst and nitrogen atom provider.

Herein, we report the extension of this methodology, using 4-hydroxycoumarin **1** as a methylene active compound.⁵ 4-Hydroxycoumarin is a recurring structural motif in many natural products of interest in medicinal chemistry.⁶

2. Results and discussion

Initially, we performed a preliminary experiment using our previously reported procedure.⁴ In refluxing EtOH (Scheme 1, path i), the addition of **1** to benzaldehyde **2a** and phenacyl chloride **3** (1:2:1) in the presence of an excess of AcOH/AcONH₄ did not yield the expected spiro compound **6**, but gave instead an already known and synthetically uninteresting mixture of products including furocoumarin **7**,^{3a} fused aziridine **8**⁷ and bisderivative **9**.⁸

In contrast, when the reaction was performed in refluxing *n*-PrOH, the addition of **1** to arylaldehydes $2\mathbf{a}-\mathbf{i}$ and **3** afforded dihydrofurocoumarins **10a**- \mathbf{i} (Scheme 1, path ii) with complete regio and diastereoselectivity in fair to good yields (47–80%) (Table 1, method A).¹⁰

The exclusive formation of the unexpected derivatives **10** prompted our interest in developing a general route to access this valuable class of compounds.

First, it was noted that AcOH/AcONH₄ mixture acted only as a basic catalyst, unlike in our previous studies.⁴ Therefore, our attention turned towards other catalysts to allow for a more efficient and selective reaction.

Using pyridine instead of the AcOH/AcONH₄ mixture this threecomponent condensation satisfactorily proceeded with a significant increase in the yield. Thus, 4-hydroxycoumarin **1** was tested with a large variety of aldehydes **2a**–**k** and α -chloroketones **3**, **4** and **5** (Scheme 1, path iii). The results are summarized in Table 1 (Method B).¹¹

The structures of new compounds **10–12** were assigned on the basis of comprehensive spectroscopic analyses and, in the case of **10a–d**, confirmed by comparison with authentic samples.¹²

The proposed mechanism for this one-pot three-component tandem Knoevenagel cyclocondensation reaction is summarized in Scheme 2. The initial regiospecific formation of the non-isolated arylidene derivative **13** is followed by its irreversible capture by the in situ-generated dipole **14** to give **15**. This Michael adduct, in contrast with our earlier report,⁸ irreversibly evolves to fused derivatives **10–12** by a regio- and stereocontrolled intramolecular five-membered ring O-cyclization. The expected trans-stereochemistry



^{*} Corresponding author. Fax: +39 090 393897; e-mail address: frisitano@ unime.it (F. Risitano).

^{0040-4020/\$ –} see front matter @ 2010 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2010.10.023



Scheme 1. Reagents and conditions: (i) 2a and 3 in AcOH/AcONH₄, EtOH, 4 Å molecular sieves, reflux, 2 h; (ii) 2a–i and 3 in AcOH/AcONH₄, *n*-PrOH, 4 Å molecular sieves, reflux, 3 h; (iii) 2a–k and 3–5 in Pyridine, *n*-PrOH, 4 Å molecular sieves, reflux, 2–6 h.

Table 1

Synthesis of 2,3-dihydrofuro[3,2-c]coumarins

R		10		11	12
		(R ¹ =Ph) yield ^a %		(R ¹ =Me) yield ^a %	$(R^1\!\!=\!\!OC_3H_7) \text{ yield}^a\%$
		Method A ^b	Method B ^c	Method B ^c	Method B ^c
a	C ₆ H ₅	80	95	85	80
b	4-MeC ₆ H ₄	78	98	85	85
с	4-MeOC ₆ H ₄	77	96		
d	4-ClC ₆ H ₄	74	95		
e	3-MeC ₆ H ₄	51	95	75	70
f	3-MeOC ₆ H ₄	47	96		
g	2-MeC ₆ H ₄	69	96	70	65
h	2-MeOC ₆ H ₄	71	94		
i	2-ClC ₆ H ₄	66	93		
j	C6H ₅ CH ₂		67		
k	CH ₃ CH ₂		64		

^a Yield of pure isolated product.

^b AcOH/AcONH₄, *n*-PrOH, 4 Å molecular sieves, reflux, 3 h.

^c Pyridine, *n*-PrOH, 4 Å molecular sieves, reflux, 2–6 h.

of the reaction final step was proved from NOE NMR studies (no interactions between H₂ and H₃) and by analysis of vicinal coupling constants of the two-methine hydrogens. Our results demonstrate that the reaction conditions are critical for the successful and exclusive formation of dihydrofurocoumarin derivatives in a diastereoselective fashion. Indeed, using method B all the electrophilic halides **3–5** are converted to ylide **14** removing possible competitive reactions.



Scheme 2. Proposed mechanism of 2,3-dihydrofuro[3,2-c]coumarin formation.

For example, with DBU the reaction of **1**, **2a** and α -chloroketones **3**, **4** or **5** stopped (Scheme 3, path iv) at the exclusive formation of the *O*-alkyl derivatives **16**, **17** or **18**, respectively.

$$1 + 2a + 3-5 \xrightarrow{(iv)} 0 \xrightarrow{0} 0 \xrightarrow{R^1} 16; R^1=Ph 37\% \\ 1000 16; R^1=Ph 37\% \\ 17; R^1=Me 40\% \\ 18; R^1=OEt 30\%$$

Scheme 3. Reaction conditions: (iv) DBU, n-PrOH, 4 Å molecular sieves, reflux, 3 h.

These structures were assigned on the basis of analytical and spectroscopic data and were confirmed by a gHMBCAD experiment on **16**. Surprisingly, these *O*-alkyl derivatives remain practically unaltered either in the presence of an aromatic aldehyde or under the influence of an acid or a base.

3. Conclusions

In summary, we have successfully prepared a series of functionalized angularly-fused dihydrofurocoumarins by an efficient multicomponent domino process. This protocol offers several advantages including high diastereoselectivity, high product yields and simple work up procedures. Further examples of the applicability of our synthetic procedures towards cyclic 1,3-dicarbonyl compounds will be reported in future publications.

4. Experimental section

4.1. General method

Melting points were determined on a Kofler melting apparatus. IR spectra were recorded in Nujol with a Nicolet Impact 410D spectrometer. ¹H and ¹³C NMR spectra were obtained on a Bruker AMX R300. The chemical shifts (δ) and coupling constants (J) are expressed in parts per million and hertz, respectively. Microanalyses were carried out on a Carlo Erba EA 1102. Merck Kieselgel 60F₂₅₄ plates were used for TLC. All solvents and reagents were obtained from commercial sources and purified before use if necessary.

4.2. General procedure for the synthesis of compounds **10**, **11**, **12**

Method A: To a stirred solution of 4-hydroxycoumarin **1** (0.5 g, 3.1 mmol) in *n*-PrOH (20 mL) containing 4 Å molecular sieves and a mixture of AcOH (10 mL) and AcONH₄ (1.55 g, 20 mmol) aldehyde **2** (6.2 mmol) and phenacyl chloride **3** (0.48 g, 3.1 mmol) were added. The mixture was heated at reflux for 3 h. After cooling, the reaction mixture was filtered to remove molecular sieves and evaporated. The resulting residue was washed with cool H₂O (20 mL) and the aqueous suspension was then extracted with CHCl₃ (3×30 mL). The combined organic layers were washed with H₂O (20 mL), dried over anhydrous Na₂SO₄, filtered and evaporated under reduced pressure. The residue was crystallized from Et₂O/acetone (9:1). Further recrystallization from MeOH led to **10a–i** as white solids.

Method B: As Method A, but using pyridine (0.47 g, 6 mml) instead of AcOH/AcONH₄ mixture, derivatives **10a**–**k** were isolated. The use of **4** (0.28 g, 3.1 mmol) or **5** (0.38 g, 3.1 mmol) in place of **3** provided **11** or **12** as white solids.

4.2.1. Compound 10a. Mp 192–194 °C. (lit.¹⁰ 193 °C).

4.2.2. Compound 10b. Mp 194–196 °C. (lit.¹⁰ 192 °C).

4.2.3. Compound 10c. Mp 177–179 °C. (lit.¹⁰ 180 °C).

4.2.4. Compound 10d. Mp 168–170 °C. (lit.¹⁰ 170–172 °C).

4.2.5. Compound **10e**. Mp 153–155 °C. IR (Nujol): 1718, 1689 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.33 (3H, s, Me); 4.74 (1H, d, CH, *J* 4.9 Hz); 6.15 (1H, d, CH, *J* 4.9 Hz); 7.07–7.91 (13H, m, CH_{arom}). ¹³C NMR (75 MHz, CDCl₃): δ 21.4; 49.2; 92.6; 105.3; 112.1; 116.8; 116.9; 123.1; 124.0; 124.6; 128.0; 128.1; 128.8; 128.9; 129.0; 129.1; 132.8; 133.0; 134.3; 138.9; 139.4; 155.3; 159.2; 166.2; 192.0. Anal. Calcd for C₂₅H₁₈O₄: C, 78.52; H, 4.74. Found: C, 78.50; H, 4.55.

4.2.6. Compound **10f**. Mp 151–152 °C. IR (Nujol): 1718, 1686 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 3.80 (3H, s, Me); 4.77 (1H, d, CH, *J* 4.9 Hz); 6.17 (1H, d, CH, *J* 4.9 Hz); 6.82–7.93 (13H, m, CH_{arom}.). ¹³C NMR (75 MHz, CDCl₃): δ 49.2; 55.2; 92.4; 105.0; 112.0; 113.2; 113.4; 116.8; 116.9; 119.7; 123.1; 140.1; 128.9; 129.0; 129.1; 130.2; 132.8; 133.0; 134.3; 141.0; 155.3; 159.2; 160.1; 166.3; 192.0. Anal. Calcd for C₂₅H₁₈O₅: C, 75.37; H, 4.55. Found: C, 75.45; H, 4.45.

4.2.7. *Compound* **10g**. Mp 201–202 °C. IR (Nujol): 1724, 1704 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.31 (3H, s, Me); 5.10 (1H, d, CH, *J* 5.0 Hz); 6.18 (1H, d, CH, *J* 5.0 Hz); 7.18–7.90 (13H, m, CH_{arom}). ¹³C NMR (75 MHz, CDCl₃): δ 19.6; 44.7; 93.0; 93.2; 106.2; 112.1; 116.8; 116.9; 122.9; 123.0; 123.9; 126.8; 126.9; 127.8; 129.0; 130.7; 132.6; 133.2; 134.3; 136.2; 138.2; 155.2; 159.2; 165.8; 192.0. Anal. Calcd for C₂₅H₁₈O₄: C, 78.52; H, 4.74. Found: C, 78.30; H, 4.70.

4.2.8. Compound **10h**. Mp 150–152 °C. IR (Nujol): 1716, 1690 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 4.10 (3H, s, Me); 4.81 (1H, d, CH, *J* 5.0 Hz); 6.11 (1H, d, CH, *J* 5.0 Hz); 7.12–7.94 (13H, m, CH_{arom}.). ¹³C NMR (75 MHz, CDCl₃): δ 44.0; 55.0; 90.7; 90.9; 103.6; 110.8; 112.2; 116.9; 120.8; 120.9; 122.8; 123.0; 123.8; 128.0; 128.6; 128.7; 128.9; 132.5; 134.0; 136.0; 138.0; 155.2; 156.8; 165.0; 192.0. Anal. Calcd for C₂₅H₁₈O₅: C, 75.37; H, 4.55. Found: C, 75.35; H, 4.75.

4.2.9. Compound **10i**. Mp 170–173 °C. IR (Nujol): 1726, 1702 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 5.47 (1H, d, CH, *J* 5.0 Hz); 6.15 (1H, d, CH, *J* 5.0 Hz); 7.23–7.98 (13H, m, CH_{arom}). ¹³C NMR (75 MHz, CDCl₃): δ 45.6; 90.9; 104.1; 112.0; 117.0; 123.0; 124.1; 125.1; 127.6; 128.5; 128.6; 128.9; 129.1; 129.3; 130.2; 131.3; 133.0; 133.5; 134.3; 136.7; 155.3; 159.2; 166.7; 191.3. Anal. Calcd for C₂₄H₁₅ClO₄: C, 71.56; H, 3.75; Cl, 8.80. Found: C, 71.60; H, 3.70; Cl, 8.82.

4.2.10. Compound **10***j*. Mp 182–184 °C. IR (Nujol): 1720, 1689 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.84 (1H, dd, $CH_{a}H_{b}Ph$, *J* 13.9, 10.3 Hz); 3.55 (1H, dd, $CH_{a}H_{b}Ph$, *J* 13.9, 3.8 Hz); 3.90 (1H, ddd, CH, *J* 13.9, 3.8, 3.4 Hz); 6.02 (1H, d, CH, *J* 3.4 Hz); 7.21–7.77 (14H, m, CH_{arom}). ¹³C NMR (75 MHz, CDCl₃): δ 38.1; 45.8; 87.4; 87.6; 104.6; 112.1; 116.8; 122.9; 124.0; 127.0; 128.4; 128.6; 128.9; 129.0; 129.5; 129.6; 132.6; 132.7; 133.7; 137.4; 155.0; 158.9; 159.9; 166.7; 192.6. Anal. Calcd for C₂₅H₁₈O₄: C, 78.52; H, 4.74. Found: C 78.55; H 4.70.

4.2.11. Compound **10k**. Mp 124–126 °C. IR (Nujol): 1718, 1704 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.02 (3H, t, Me, *J* 7.4 Hz); 1.96 (1H, sex, *CH*_aH_bMe, *J* 13.8, 7.4, 3.8 Hz); 2.06 (1H, sex,

CH_a*H*_bMe, *J* 13.8, 7.4, 3.8 Hz); 3.83 (1H, ddd, CH, *J* 6.8, 4.8, 3.8 Hz); 5.92 (1H, d, CH, *J* 4.8 Hz); 7.28−8.02 (9H, m, CH_{arom}). ¹³C NMR (75 MHz, CDCl₃): δ 10.1; 24.8; 44.5; 104.3; 112.1; 116.8; 116.9; 122.8; 122.9; 124.0; 124.1; 128.8; 128.9; 132.5; 133.7; 134.1; 155.0; 159.9; 166.1; 193.3. Anal. Calcd for C₂₀H₁₆O₄: C, 74.99; H, 5.03. Found: C, 75.00; H, 5.05.

4.2.12. Compound **11a**. Mp 127–128 °C. IR (Nujol): 1734, 1692 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.37 (1H, s, COMe); 4.76 (1H, d, CH, J 5.2 Hz); 5.23 (1H, d, CH, J 5.2 Hz); 7.26–7.85 (9H, m, CH_{arom.}). ¹³C NMR (75 MHz, CDCl₃): δ 26.1; 48.7; 95.6; 104.8; 111.8; 117.0; 122.8; 124.1; 127.1; 127.2; 127.8; 129.0; 129.1; 132.9; 139.5; 155.2; 159.0; 165.8; 203.3. Anal. Calcd for C₁₉H₁₄O₄: C, 74.50; H, 4.61. Found: C, 74.52; H, 4.62.

4.2.13. Compound **11b**. Mp 113–115 °C. IR (Nujol): 1735, 1695 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.29 (3H, s, COMe); 2.32 (3H, s, Me); 4.69 (1H, d, CH, *J* 5.2 Hz); 5.20 (1H, d, CH, *J* 5.2 Hz); 7.03–7.82 (8H, m, CH_{arom}.). ¹³C NMR (75 MHz, CDCl₃): δ 20.9; 26.0; 48.4; 95.6; 104.8; 111.8; 116.9; 122.7; 124.0; 126.9; 127.0; 129.6; 129.7; 132.8; 136.5; 137.4; 155.1; 158.9; 165.7; 203.3. Anal. Calcd for C₂₀H₁₆O₄: C, 74.99; H 5.03. Found: C, 75.01; H, 5.06.

4.2.14. Compound **11e**. Mp 104–106 °C. IR (Nujol): 1733, 1700 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.29 (3H, s, COMe); 2.32 (3H, s, Me); 4.69 (1H, d, CH, *J* 5.2 Hz); 5.20 (1H, d, CH, *J* 5.2 Hz); 7.03–7.82 (8H, m, CH_{arom}). ¹³C NMR (75 MHz, CDCl₃): δ 21.1; 25.9; 48.4; 95.4; 104.6; 111.6; 116.7; 122.6; 123.9; 127.6; 128.4; 128.5; 128.7; 132.7; 138.5; 139.4; 154.9; 158.8; 165.6; 203.4. Anal. Calcd for C₂₀H₁₆O₄: C, 74.99; H, 5.03. Found: C, 74.98; H, 5.05.

4.2.15. Compound **11g**. Mp 157–159 °C. IR (Nujol): 1734, 1699 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.36 (3H, s, COMe); 2.54 (3H, s, Me); 5.02 (1H, d, CH, J 4.6 Hz); 5.20 (1H, d, CH, J 4.6 Hz); 7.01–7.85 (8H, m, CH_{arom.}). ¹³C NMR (75 MHz, CDCl₃): δ 19.7; 26.1; 44.4; 95.9; 105.9; 111.9; 117.0; 122.6; 124.1; 126.4; 126.7; 127.5; 130.8; 132.8; 136.0; 137.9; 155.1; 158.8; 165.4; 203.2. Anal. Calcd for C₂₀H₁₆O₄: C, 74.99; H, 5.03. Found: C, 75.02; H, 5.05.

4.2.16. Compound **12a**. Mp 101–102 °C. IR (Nujol): 1729, 1717 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.34 (3H, t, CH₂*Me*, *J* 7.0 Hz); 1.70–1.73 (2H, m, *CH*₂*Me*); 4.25–4.27 (2H, m, OCH₂); 4.78 (d, CH, *J* 5.0 Hz); 5.28 (d, CH, *J* 5.0 Hz); 7.28–7.85 (9H, m, CH_{arom}.). ¹³C NMR (75 MHz, CDCl₃): δ 13.9; 21.6; 50.2; 62.1; 88.9; 104.4; 111.8; 116.8; 123.0; 124.0; 126.8; 126.9; 127.8; 128.8; 128.9; 132.8; 139.3; 155.1; 159.1; 166.4; 168.5. Anal. Calcd for C₂₁H₁₈O₅: C, 71.99; H, 5.18. Found: C, 71.98; H, 5.20.

4.2.17. Compound **12b**. Mp 132–133 °C. IR (Nujol): 1730, 1718 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.31 (3H, t, CH₂*Me*, *J* 7.2 Hz); 1.80–1.84 (2H, m, *CH*₂*Me*); 2.33 (3H, s, Me); 4.33–4.37 (2H, m, OCH₂); 4.74 (1H, d, CH, *J* 5.0 Hz); 5.25 (1H, d, CH, *J* 5.0 Hz); 7.17–7.85 (8H, m, CH_{arom.}). ¹³C NMR (75 MHz, CDCl₃): δ 14.1; 21.0; 21.2; 50.1; 62.2; 89.2; 104.7; 12.0; 116.9; 123.1; 124.1; 126.8; 126.9; 129.7; 129.8; 132.8; 136.5; 137.7; 155.2; 159.2; 166.4; 168.7. Anal. Calcd for C₂₂H₂₀O₅: C, 72.51; H, 5.53. Found: C, 72.52; H, 5.55.

4.2.18. Compound **12e**. Mp 100–101 °C. IR (Nujol): 1755, 1719 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.35 (3H, t, CH₂*Me*, *J* 7.1 Hz); 1.75–1.79 (2H, m, CH₂Me); 2.33 (3H, s, Me); 4.32–4.36 (2H, m, OCH₂); 4.73 (1H, d, CH, *J* 4.6 Hz); 5.27 (1H, d, CH, *J* 4.6 Hz); 7.07–7.85 (8H, m, CH_{arom}). ¹³C NMR (75 MHz, CDCl₃): δ 14.1; 21.4; 21.6; 50.3; 62.2; 89.2; 104.7; 112.0; 117.0; 123.1; 124.1; 124.2; 127.7; 128.8; 129.0; 132.9; 138.8; 139.4; 155.3; 159.2; 166.4; 168.7. Anal. Calcd for C₂₂H₂₀O₅: C, 72.51; H, 5.53. Found: C, 72.53; H, 5.56.

4.2.19. Compound **12g.** Mp 137–138 °C. IR (Nujol): 1752, 1720 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.32 (3H, t, CH₂Me, *J* 7.0 Hz); 1.72–1.76 (2H, m, *CH*₂Me); 2.23 (3H, s, Me); 4.20–4.24 (2H, m, OCH₂); 5.00 (1H, d, CH, *J* 4.4 Hz); 5.23 (1H, d, CH, *J* 4.4 Hz); 6.99–7.85 (8H, m, CH_{arom.}). ¹³C NMR (75 MHz, CDCl₃): δ 14.1; 19.7; 21.2; 46.2; 62.2; 89.2; 104.8; 114.0; 117.0; 123.0; 124.1; 126.2; 126.8; 127.8; 130.9; 132.8; 138.8; 139.4; 155.3; 159.2; 166.4; 168.7. Anal. Calcd for C₂₂H₂₀O₅: C, 72.51; H, 5.53. Found: C, 72.52; H, 5.54.

4.3. General procedure for the synthesis of compounds **16**, **17**, **18**

To a stirred solution of 4-hydroxycoumarin **1** (0.5 g, 3.1 mmol) in *n*-PrOH (20 mL), containing 4 Å molecular sieves and an excess of DBU (0.56 g, 3.7 mmol), benzaldehyde **2a** (0.66 g, 6.2 mmol) and α -chloroketones **3**, **4** or **5** (3.1 mmol) were added. The resulting mixture was heated at reflux for 3 h. After cooling, the reaction products **16**, **17** and **18**, respectively, were collected for filtration. Recrystallization from MeOH led to colourless needles. Benzalde-hyde **2a** was recovered unaltered.

4.3.1. Compound **16**. Mp 190–192 °C. IR (Nujol): 1701, 1713 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 5.48 (2H, s, OCH₂), 5.59 (1H, s, CH), 7.26–7.99 (9H, m, CH_{arom}). ¹³C NMR (75 MHz, CDCl₃): δ 70.4; 91.3; 115.0; 116.7; 123.8; 124.0; 127.8; 127.9; 129.0; 129.1; 132.6; 134.0; 134.4; 153.3; 162.4; 164.9; 184.4. Anal. Calcd for C₁₇H₁₂O₄: C, 72.85; H, 4.32. Found: C, 72.75; H, 4.30.

4.3.2. Compound **17**. Mp 173–175 °C. IR (Nujol): 1699, 1710 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.36 (3H, s, COMe), 4.75 (2H, s, OCH₂), 5.55 (1H, s, CH), 7.29–7.93 (4H, m, CH_{arom.}). ¹³C NMR (75 MHz, CDCl₃): δ 26.5; 72.5; 91.2; 115.0; 116.7; 123.0; 124.0; 132.7; 153.2; 162.2; 164.4; 200.8. Anal. Calcd for C₁₂H₁₀O₄: C, 66.05; H, 4.62. Found: C, 66.25; H, 4.49.

4.3.3. *Compound* **18**. Mp 86–88 °C. IR (Nujol): 1712, 1745 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 0.95 (3H, t, CH₂*Me*, *J* 7.45 Hz), 1.70–1.74 (2H, m, *CH*₂Me, *J* 7.45, 6.85 Hz), 4.20 (2H, t, *CH*₂CH₂Me, *J* 6.85 Hz), 4.77 (2H, s, OCH₂), 5.57 (1H, s, CH), 7.26–7.92 (4H, m, CH_{arom.}). ¹³C NMR (75 MHz, CDCl₃): δ 10.1; 21.7; 65.2; 67.4; 91.0; 115.1; 116.7; 123.2; 123.9; 132.7; 153.2; 162.3; 164.6; 166.4. Anal. Calcd for C₁₄H₁₄O₅: C, 64.12; H, 5.38. Found: C, 64.10; H, 5.33.

Supplementary data

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.tet.2010.10.023.

References and notes

- For reviewes, see: (a) Ho, T. L. Tandem Organic Reactions; Wiley: New York, NY, 1992; (b) Denmark, S. E.; Thorarensen, A. Chem. Rev. **1996**, *96*, 137–166; (c) Tietze, L. F. Chem. Rev. **1996**, 96, 115–136; (d) Dömling, A.; Ugi, I. Angew. Chem., Int. Ed. **2000**, 39, 3168–3210; (e) Bienaymè, H.; Hulme, C.; Oddon, G.; Schmitt, P. Chem.—Eur. J. **2000**, 6, 3321–3329.
- (a) Rieger, D.; Lotz, S. D.; Kernbach, U.; Andrè, C.; Bertran-Nadal, J.; Fehlammer, W. P. J. Organomet. Chem. **1995**, 491, 135–152; (b) Armstrong, R. W.; Combs, A. P.; Tempest, P. A.; Brown, S. D.; Keating, T. A. Acc. Chem. Res. **1996**, 29, 123–131; (c) Nefzi, A.; Ostresh, J. M.; Houghten, R. A. Chem. Rev. **1997**, 97, 449–472; (d) Weber, L; Illgen, K.; Almstetter, M. Synlett **1999**, 366–374.
- (a) Risitano, F.; Grassi, G.; Foti, F. J. Heterocycl. Chem. 2001, 38, 1083–1089; (b) Risitano, F.; Grassi, G.; Foti, F.; Biliardo, C. Tetrahedron Lett. 2001, 42, 3503–3505; (c) Risitano, F.; Grassi, G.; Foti, F.; Nicolò, F.; Condello, M. Tetrahedron 2002, 58, 191–195; (d) Risitano, F.; Grassi, G.; Foti, F.; Romeo, R. Synthesis 2002, 116–120; (e) Grassi, G.; Risitano, F.; Foti, F.; Cordaro, M.; Bruno, G.; Nicolò, F. Chem. Commun. 2003, 1868–1869.
- Altieri, E.; Cordaro, M.; Grassi, G.; Risitano, F.; Scala, A. Synlett 2010, 14, 2106–2108.
- Risitano F. Grassi G. Cordaro M. Liotta C. Moraci S. Abstracts of Papers, 3rd International Conference on Multi-Component Reactions and Related Chemistry; RSC-Advancing the Chemical Sciences: Amsterdam, The Netherlands, July 9–13, 2006.
- (a) Murray, R. D. H.; Mendez, J.; Brown, S. A. The Natural Coumarins, Occurrence, Chemistry, and Biochemistry; Wiley: New York, NY, 1982; 13; (b) Gunatilaka, A. A. L.; Kingston, D. G. I.; Wijeratne, E. M. K.; Bandara, B. M. R.; Hofmann, G. A.; Johnson, R. K. J. Nat. Prod. **1994**, 57, 518–520; (c) O'Kennedy, R.; Thornes, R. D. Coumarins: Biology, Applications and Mode of Action; Wiley: Chichester, 1997; (d) Takeuchi, Y.; Xie, L.; Cosentino, L. M.; Lee, K. H. Bioorg. Med. Chem. Lett. **1997**, 7, 2573–2578; (e) Zhao, H.; Neamati, N.; Hong, H.; Mazumder, A.; Wang, S.; Sunder, S.; Milne, G. W. A.; Pommier, Y.; Burke, T. R., Jr. J. Med. Chem. **1997**, 40, 242–449; (f) Magiatis, P.; Melliou, E.; Skaltsounis, A.; Mitaku, S.; Leonce, S.; Renard, P.; Pierre, A.; Alassi, G. J. Nat. Prod. **1998**, 61, 982–986; (g) Asomaning, W. A.; Otoo, E.; Akoto, O.; Oppong, I. V.; Addae-Mensah, I.; Waibel, R.; Achenbach, H. Phytochemistry **1999**, 51, 937–941.
- (a) Risitano, F.; Grassi, G.; Foti, F.; Moraci, S. Synlett 2005, 1633–1635; (b) Bruno,
 G.; Rotondo, A.; Nicolò, F.; Risitano, F.; Grassi, G.; Foti, F. Helv. Chim. Acta 2006,
 89, 190–200.
- 8. Risitano, F.; Grassi, G.; Bruno, G.; Nicolò, F. Liebigs Ann. 1997, 441-445.
- Performing the reaction with 5 (R¹=OCH₂CH₃) in *n*-PrOH, an unavoidable transesterification led to exclusive formation of 12 (R¹=OCH₂CH₂CH₃).
- 10. Under these conditions, no reaction occurred with aliphatic aldehydes 2j, k or with aliphatic α -halogen ketones 4 and 5.
- Our one-pot procedure is more convenient with respect to Ref.12 in terms of yield and reaction time.
- 12. Wang, Q. F.; Hou, H.; Hui, L.; Yan, C. C. J. Org. Chem. 2009, 74, 7403-7406.